

Diethyl 2,2'-[(5-dimethylamino-1-naphthyl)sulfonylimino]diacetate

Yong Zhang,* Yuan Qu and Ting Liu

School of Chemical and Materials Engineering, Huangshi Institute of Technology, Huangshi 435003, People's Republic of China
 Correspondence e-mail: zy0340907@yahoo.com.cn

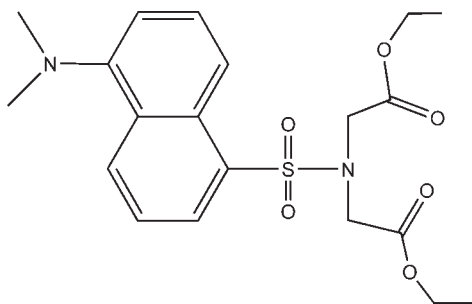
Received 23 September 2009; accepted 11 October 2009

Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; disorder in main residue; R factor = 0.053; wR factor = 0.145; data-to-parameter ratio = 13.7.

In the title compound, $\text{C}_{20}\text{H}_{26}\text{N}_2\text{O}_6\text{S}$, the N atom of the dimethylamino group is displaced by 0.113 (2) Å from the plane of the naphthalene ring system. The two ethoxy groups adopt zigzag conformations. In the crystal structure, weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules, forming a three-dimensional network. Both ethyl groups are disordered over two sites with the ratios of refined occupancies being 0.857 (16):0.143 (16) and 0.517 (14):0.483 (14).

Related literature

For applications of ligands containing the 5-(dimethylamino) naphthalene-1-sulfonyl (dansyl) group, see: Corradini *et al.* (1997); Christoforou *et al.* (2006); Zhang *et al.* (2009).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{26}\text{N}_2\text{O}_6\text{S}$
 $M_r = 422.49$
 Monoclinic, $P2_1/c$
 $a = 13.1266$ (9) Å
 $b = 8.4592$ (5) Å
 $c = 19.3206$ (12) Å
 $\beta = 93.530$ (1)°

$V = 2141.3$ (2) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.19$ mm⁻¹
 $T = 298$ K
 $0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker SMART CCD diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.970$, $T_{\max} = 0.981$

16359 measured reflections
 4201 independent reflections
 3705 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.040$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.053$
 $wR(F^2) = 0.145$
 $S = 1.05$
 4201 reflections
 306 parameters

12 restraints
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.35$ e Å⁻³
 $\Delta\rho_{\min} = -0.27$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C4}-\text{H4}\cdots\text{O1}$	0.93	2.36	3.006 (3)	126
$\text{C13}-\text{H13A}\cdots\text{O2}^i$	0.97	2.36	3.272 (3)	157

 Symmetry code: (i) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: SMART (Bruker, 2007); cell refinement: SAINT-Plus (Bruker, 2007); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009); software used to prepare material for publication: SHELXTL (Sheldrick, 2008).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2912).

References

- Bruker (2007). SAINT-Plus and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
 Christoforou, A. M., Marzilli, P. A. & Marzilli, L. G. (2006). *Inorg. Chem.* **45**, 6771–6781.
 Corradini, R., Dossena, A., Galaverna, G., Marchelli, R., Panagia, A. & Sarto, G. (1997). *J. Org. Chem.* **62**, 6283–6289.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
 Zhang, S., Zhao, B., Su, Z., Xia, X. & Zhang, Y. (2009). *Acta Cryst.* **E65**, o1452.

supplementary materials

Acta Cryst. (2009). E65, o2752 [doi:10.1107/S1600536809041476]

Diethyl 2,2'-[(5-dimethylamino-1-naphthyl)sulfonylimino]diacetate

Y. Zhang, Y. Qu and T. Liu

Comment

Dansyl (5-(dimethylamino)naphthalene-1-sulfonyl) derivatives are of considerable interest because of their good fluorescent properties. Many fluorescent ligands bearing dansyl group have been reported in recent years (e.g. Corradini *et al.*, 1997; Christoforou *et al.*, 2006; Zhang *et al.*, 2009). We are interested in preparing fluorescent ligands that are expected to bind to hydrophobic sites in proteins or membranes. With this in mind, the title compound, (I), was prepared and we report herein the crystal structure.

In the molecular structure (Fig. 1), the N atom of the dimethylamino group is displaced by 0.113 (2) Å from the plane of the naphthalene ring system. The two ethoxycarbonyl groups adopt coiled conformations with C14—O4—C15—C16 and C18—O6—C119—C20 torsion angles of 90.1 (10)° and 159.6 (9)°, respectively. All bond lengths and bond angles are as expected. In the crystal structure (Fig.2) weak intermolecular C—H···O hydrogen bonds lead to the formation of a three-dimension network.

Experimental

Diethyl iminodiacetate(0.38 g, 2 mmol) was added to a stirred solution of dansyl chloride(0.27 g, 1 mmol) in dry acetonitrile(40 ml). The reaction mixture was allowed to stir for 12 hr at 353 K. The progress of the reaction was monitored by TLC, untill the completion of reaction. The solvent was evaporated and the residue was purified by column chromatography (hexane-ethyl acetate,1:5 v/v) to afford the title compound as a yellow solid. Single crystals suitable for X-ray diffraction were prepared by slow evaporation of a solution of the title compound in ethyl acetate at room temperature.

Refinement

Both the C15/C16 and C18/C19 ethyl groups are disordered over two positions with the final refined occupancies of 0.857 (16):0.143 (16) and 0.517 (14):0.483 (14) for the major and minor components, respectively. In the refinement, commands 'DFIX' and 'SADI' were used (Sheldrick, 2008). Some B-level alert using *PLATON* were resulted from the disorder.

All H atoms were placed in idealized positions [C—H=0.96 Å (methyl), 0.97Å (methylene) and 0.93 Å (aromatic)] and included in the refinement in the riding-model approximation, with $U_{\text{iso}}(\text{H})=1.5U_{\text{eq}}(\text{methyl C})$ and $1.2U_{\text{eq}}(\text{methylene and aromatic C})$.

Figures

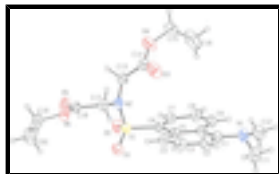


Fig. 1. The molecular structure of (I), with displacement ellipsoids drawn at the 50% probability level. The disorder is not shown.

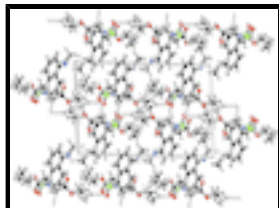


Fig. 2. Part of the crystal structure of (I) showing weak hydrogen bonds as dashed lines.

Diethyl 2,2'-[(5-dimethylamino-1-naphthyl)sulfonylimino]diacetate

Crystal data

$C_{20}H_{26}N_2O_6S$

$M_r = 422.49$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 13.1266$ (9) Å

$b = 8.4592$ (5) Å

$c = 19.3206$ (12) Å

$\beta = 93.530$ (1)°

$V = 2141.3$ (2) Å³

$Z = 4$

$F_{000} = 896$

$D_x = 1.311$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 7099 reflections

$\theta = 2.4$ – 27.8 °

$\mu = 0.19$ mm⁻¹

$T = 298$ K

Block, yellow

$0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker SMART CCD
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 298$ K

φ and ω scans

Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)

$T_{\min} = 0.970$, $T_{\max} = 0.981$

16359 measured reflections

4201 independent reflections

3705 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.040$

$\theta_{\text{max}} = 26.0$ °

$\theta_{\text{min}} = 2.5$ °

$h = -15 \rightarrow 16$

$k = -10 \rightarrow 10$

$l = -23 \rightarrow 23$

Refinement

Refinement on F^2

Least-squares matrix: full

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

$$R[F^2 > 2\sigma(F^2)] = 0.053$$

$$wR(F^2) = 0.145$$

$$S = 1.05$$

4201 reflections

306 parameters

12 restraints

Primary atom site location: structure-invariant direct methods

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0753P)^2 + 0.7093P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.35 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.27 \text{ e } \text{\AA}^{-3}$$

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1	0.95264 (17)	0.5912 (3)	0.37909 (11)	0.0578 (5)	
C2	1.02352 (18)	0.4871 (3)	0.35796 (14)	0.0715 (7)	
H2	1.0911	0.4969	0.3750	0.086*	
C3	0.99669 (18)	0.3659 (3)	0.31122 (15)	0.0737 (7)	
H3	1.0466	0.2959	0.2981	0.088*	
C4	0.89929 (16)	0.3482 (3)	0.28453 (13)	0.0622 (6)	
H4	0.8836	0.2689	0.2523	0.075*	
C5	0.82149 (15)	0.4499 (2)	0.30558 (10)	0.0466 (4)	
C6	0.71621 (15)	0.4373 (2)	0.28232 (10)	0.0446 (4)	
C7	0.64297 (16)	0.5326 (2)	0.30780 (11)	0.0522 (5)	
H7	0.5752	0.5227	0.2912	0.063*	
C8	0.67007 (17)	0.6443 (3)	0.35867 (12)	0.0588 (5)	
H8	0.6200	0.7065	0.3771	0.071*	
C9	0.76932 (17)	0.6628 (3)	0.38138 (12)	0.0575 (5)	
H9	0.7861	0.7382	0.4153	0.069*	
C10	0.84776 (15)	0.5708 (2)	0.35492 (10)	0.0490 (5)	
C11	0.9684 (3)	0.8725 (4)	0.39713 (18)	0.0941 (10)	
H11A	1.0269	0.8936	0.3712	0.141*	
H11B	0.9651	0.9487	0.4337	0.141*	
H11C	0.9076	0.8794	0.3669	0.141*	
C12	1.0732 (2)	0.6972 (5)	0.46859 (16)	0.0957 (10)	
H12A	1.0791	0.5906	0.4854	0.144*	
H12B	1.0742	0.7688	0.5072	0.144*	

supplementary materials

H12C	1.1293	0.7203	0.4406	0.144*	
C13	0.59995 (17)	0.1130 (3)	0.32052 (11)	0.0548 (5)	
H13A	0.5668	0.0108	0.3159	0.066*	
H13B	0.5478	0.1937	0.3136	0.066*	
C14	0.64919 (19)	0.1283 (3)	0.39272 (12)	0.0588 (5)	
C15	0.6061 (8)	0.1733 (7)	0.5102 (2)	0.0874 (19)	0.857 (16)
H15A	0.6740	0.1337	0.5230	0.105*	0.857 (16)
H15B	0.5583	0.1241	0.5399	0.105*	0.857 (16)
C16	0.6032 (10)	0.3472 (8)	0.5176 (3)	0.149 (4)	0.857 (16)
H16A	0.6565	0.3935	0.4925	0.223*	0.857 (16)
H16B	0.6128	0.3748	0.5658	0.223*	0.857 (16)
H16C	0.5383	0.3861	0.4994	0.223*	0.857 (16)
C17	0.73411 (17)	-0.0060 (3)	0.24933 (12)	0.0552 (5)	
H17A	0.7484	-0.0699	0.2905	0.066*	
H17B	0.7988	0.0315	0.2339	0.066*	
C18	0.68302 (16)	-0.1075 (2)	0.19336 (11)	0.0520 (5)	
C19	0.7087 (7)	-0.2942 (12)	0.1072 (5)	0.073 (3)	0.517 (14)
H19A	0.6722	-0.2204	0.0762	0.088*	0.517 (14)
H19B	0.6624	-0.3784	0.1185	0.088*	0.517 (14)
C20	0.8001 (6)	-0.3609 (15)	0.0734 (5)	0.100 (3)	0.517 (14)
H20A	0.8432	-0.2759	0.0600	0.150*	0.517 (14)
H20B	0.7776	-0.4207	0.0331	0.150*	0.517 (14)
H20C	0.8378	-0.4285	0.1057	0.150*	0.517 (14)
C15'	0.626 (2)	0.223 (5)	0.4978 (12)	0.069 (9)	0.143 (16)
H15C	0.6659	0.3117	0.4825	0.083*	0.143 (16)
H15D	0.6711	0.1531	0.5253	0.083*	0.143 (16)
C16'	0.539 (3)	0.281 (6)	0.5403 (16)	0.122 (14)	0.143 (16)
H16D	0.5088	0.3735	0.5189	0.182*	0.143 (16)
H16E	0.5652	0.3060	0.5864	0.182*	0.143 (16)
H16F	0.4883	0.1993	0.5421	0.182*	0.143 (16)
C19'	0.7221 (11)	-0.3420 (9)	0.1268 (4)	0.069 (3)	0.483 (14)
H19C	0.6507	-0.3697	0.1290	0.083*	0.483 (14)
H19D	0.7638	-0.4342	0.1382	0.083*	0.483 (14)
C20'	0.7438 (12)	-0.2768 (12)	0.0564 (4)	0.101 (4)	0.483 (14)
H20D	0.7039	-0.1831	0.0474	0.152*	0.483 (14)
H20E	0.7263	-0.3544	0.0215	0.152*	0.483 (14)
H20F	0.8150	-0.2515	0.0556	0.152*	0.483 (14)
N1	0.97658 (16)	0.7153 (3)	0.42652 (11)	0.0706 (6)	
N2	0.67264 (13)	0.1287 (2)	0.26716 (9)	0.0524 (4)	
O1	0.74296 (13)	0.2660 (2)	0.16995 (8)	0.0631 (4)	
O2	0.56924 (12)	0.3303 (2)	0.20057 (8)	0.0635 (4)	
O3	0.73809 (15)	0.1308 (3)	0.40732 (10)	0.0988 (7)	
O4	0.57813 (15)	0.1376 (3)	0.43757 (9)	0.0934 (7)	
O5	0.59653 (12)	-0.0973 (2)	0.17180 (10)	0.0751 (5)	
O6	0.74949 (13)	-0.2111 (2)	0.17216 (11)	0.0783 (6)	
S1	0.67202 (4)	0.28980 (6)	0.22222 (3)	0.04759 (18)	

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0469 (12)	0.0728 (14)	0.0531 (12)	−0.0099 (10)	−0.0019 (9)	0.0004 (10)
C2	0.0377 (12)	0.0952 (19)	0.0807 (17)	−0.0057 (12)	−0.0047 (11)	−0.0049 (14)
C3	0.0409 (12)	0.0834 (18)	0.0969 (19)	0.0088 (12)	0.0039 (12)	−0.0136 (15)
C4	0.0439 (12)	0.0645 (13)	0.0783 (15)	0.0032 (10)	0.0028 (10)	−0.0126 (12)
C5	0.0390 (10)	0.0492 (10)	0.0516 (11)	−0.0017 (8)	0.0024 (8)	0.0026 (9)
C6	0.0416 (10)	0.0439 (10)	0.0479 (10)	−0.0013 (8)	−0.0012 (8)	0.0002 (8)
C7	0.0409 (11)	0.0525 (11)	0.0625 (12)	0.0024 (9)	−0.0020 (9)	−0.0013 (10)
C8	0.0492 (12)	0.0585 (12)	0.0691 (14)	0.0070 (10)	0.0073 (10)	−0.0115 (11)
C9	0.0574 (13)	0.0585 (12)	0.0567 (12)	−0.0042 (10)	0.0027 (10)	−0.0115 (10)
C10	0.0433 (11)	0.0553 (11)	0.0481 (11)	−0.0033 (9)	−0.0003 (8)	0.0032 (9)
C11	0.094 (2)	0.086 (2)	0.100 (2)	−0.0227 (17)	−0.0109 (18)	−0.0207 (18)
C12	0.0681 (18)	0.139 (3)	0.0770 (18)	−0.0235 (18)	−0.0197 (14)	−0.0136 (18)
C13	0.0504 (12)	0.0551 (12)	0.0586 (12)	−0.0068 (9)	0.0010 (9)	−0.0018 (10)
C14	0.0591 (14)	0.0581 (13)	0.0587 (13)	0.0007 (11)	0.0002 (11)	−0.0021 (10)
C15	0.106 (4)	0.108 (4)	0.048 (2)	0.013 (3)	0.008 (2)	0.010 (2)
C16	0.252 (12)	0.115 (4)	0.075 (3)	0.068 (5)	−0.026 (4)	−0.014 (3)
C17	0.0509 (12)	0.0492 (11)	0.0642 (13)	0.0078 (9)	−0.0064 (10)	−0.0050 (9)
C18	0.0450 (12)	0.0507 (11)	0.0612 (12)	−0.0067 (9)	0.0104 (9)	−0.0046 (9)
C19	0.070 (4)	0.055 (5)	0.093 (7)	0.002 (4)	−0.003 (5)	−0.027 (5)
C20	0.091 (5)	0.108 (7)	0.103 (5)	0.003 (4)	0.019 (4)	−0.038 (5)
C15'	0.078 (16)	0.09 (2)	0.044 (12)	0.002 (13)	0.007 (11)	0.002 (13)
C16'	0.11 (3)	0.13 (3)	0.12 (2)	0.04 (2)	−0.006 (19)	0.01 (2)
C19'	0.092 (7)	0.044 (4)	0.074 (4)	−0.011 (4)	0.018 (4)	−0.005 (3)
C20'	0.153 (11)	0.085 (6)	0.067 (4)	−0.032 (6)	0.025 (5)	−0.010 (4)
N1	0.0575 (12)	0.0899 (16)	0.0630 (12)	−0.0152 (11)	−0.0086 (9)	−0.0105 (11)
N2	0.0545 (10)	0.0458 (9)	0.0570 (10)	0.0032 (8)	0.0041 (8)	−0.0042 (8)
O1	0.0666 (10)	0.0730 (10)	0.0499 (8)	0.0011 (8)	0.0046 (7)	−0.0043 (7)
O2	0.0521 (9)	0.0667 (9)	0.0688 (10)	0.0086 (7)	−0.0194 (7)	−0.0052 (8)
O3	0.0622 (12)	0.161 (2)	0.0711 (12)	0.0146 (13)	−0.0128 (9)	−0.0102 (13)
O4	0.0767 (13)	0.1434 (19)	0.0607 (11)	−0.0036 (13)	0.0095 (9)	−0.0127 (12)
O5	0.0455 (9)	0.0878 (12)	0.0910 (12)	−0.0035 (8)	−0.0051 (8)	−0.0257 (10)
O6	0.0566 (10)	0.0703 (11)	0.1090 (14)	−0.0017 (8)	0.0126 (9)	−0.0396 (10)
S1	0.0445 (3)	0.0501 (3)	0.0474 (3)	0.0023 (2)	−0.0045 (2)	−0.0020 (2)

Geometric parameters (Å, °)

C1—C2	1.362 (4)	C15—H15A	0.9700
C1—N1	1.416 (3)	C15—H15B	0.9700
C1—C10	1.437 (3)	C16—H16A	0.9600
C2—C3	1.397 (4)	C16—H16B	0.9600
C2—H2	0.9300	C16—H16C	0.9600
C3—C4	1.357 (3)	C17—N2	1.450 (3)
C3—H3	0.9300	C17—C18	1.506 (3)
C4—C5	1.414 (3)	C17—H17A	0.9700
C4—H4	0.9300	C17—H17B	0.9700

supplementary materials

C5—C10	1.426 (3)	C18—O5	1.188 (3)
C5—C6	1.431 (3)	C18—O6	1.319 (3)
C6—C7	1.369 (3)	C19—O6	1.509 (7)
C6—S1	1.777 (2)	C19—C20	1.510 (8)
C7—C8	1.394 (3)	C19—H19A	0.9700
C7—H7	0.9300	C19—H19B	0.9700
C8—C9	1.358 (3)	C20—H20A	0.9600
C8—H8	0.9300	C20—H20B	0.9600
C9—C10	1.412 (3)	C20—H20C	0.9600
C9—H9	0.9300	C15'—O4	1.477 (10)
C11—N1	1.447 (4)	C15'—C16'	1.529 (10)
C11—H11A	0.9600	C15'—H15C	0.9700
C11—H11B	0.9600	C15'—H15D	0.9700
C11—H11C	0.9600	C16'—H16D	0.9600
C12—N1	1.472 (3)	C16'—H16E	0.9600
C12—H12A	0.9600	C16'—H16F	0.9600
C12—H12B	0.9600	C19'—O6	1.444 (8)
C12—H12C	0.9600	C19'—C20'	1.511 (8)
C13—N2	1.453 (3)	C19'—H19C	0.9700
C13—C14	1.506 (3)	C19'—H19D	0.9700
C13—H13A	0.9700	C20'—H20D	0.9600
C13—H13B	0.9700	C20'—H20E	0.9600
C14—O3	1.184 (3)	C20'—H20F	0.9600
C14—O4	1.314 (3)	N2—S1	1.6160 (18)
C15—O4	1.460 (5)	O1—S1	1.4295 (17)
C15—C16	1.478 (7)	O2—S1	1.4292 (15)
C2—C1—N1	123.0 (2)	H15A—C15—H15B	108.6
C2—C1—C10	118.9 (2)	N2—C17—C18	112.89 (18)
N1—C1—C10	118.0 (2)	N2—C17—H17A	109.0
C1—C2—C3	121.2 (2)	C18—C17—H17A	109.0
C1—C2—H2	119.4	N2—C17—H17B	109.0
C3—C2—H2	119.4	C18—C17—H17B	109.0
C4—C3—C2	121.4 (2)	H17A—C17—H17B	107.8
C4—C3—H3	119.3	O5—C18—O6	125.1 (2)
C2—C3—H3	119.3	O5—C18—C17	125.7 (2)
C3—C4—C5	120.2 (2)	O6—C18—C17	109.14 (19)
C3—C4—H4	119.9	O6—C19—C20	106.4 (6)
C5—C4—H4	119.9	O6—C19—H19A	110.4
C4—C5—C10	118.66 (19)	C20—C19—H19A	110.4
C4—C5—C6	124.39 (19)	O6—C19—H19B	110.4
C10—C5—C6	116.92 (18)	C20—C19—H19B	110.4
C7—C6—C5	121.96 (18)	H19A—C19—H19B	108.6
C7—C6—S1	116.10 (15)	O4—C15'—C16'	106.3 (13)
C5—C6—S1	121.85 (15)	O4—C15'—H15C	110.5
C6—C7—C8	119.89 (19)	C16'—C15'—H15C	110.5
C6—C7—H7	120.1	O4—C15'—H15D	110.5
C8—C7—H7	120.1	C16'—C15'—H15D	110.5
C9—C8—C7	120.3 (2)	H15C—C15'—H15D	108.7
C9—C8—H8	119.9	C15'—C16'—H16D	109.5

C7—C8—H8	119.9	C15'—C16'—H16E	109.5
C8—C9—C10	121.7 (2)	H16D—C16'—H16E	109.5
C8—C9—H9	119.1	C15'—C16'—H16F	109.5
C10—C9—H9	119.1	H16D—C16'—H16F	109.5
C9—C10—C5	119.07 (18)	H16E—C16'—H16F	109.5
C9—C10—C1	121.5 (2)	O6—C19'—C20'	102.3 (6)
C5—C10—C1	119.39 (19)	O6—C19'—H19C	111.3
N1—C11—H11A	109.5	C20'—C19'—H19C	111.3
N1—C11—H11B	109.5	O6—C19'—H19D	111.3
H11A—C11—H11B	109.5	C20'—C19'—H19D	111.3
N1—C11—H11C	109.5	H19C—C19'—H19D	109.2
H11A—C11—H11C	109.5	C19'—C20'—H20D	109.5
H11B—C11—H11C	109.5	C19'—C20'—H20E	109.5
N1—C12—H12A	109.5	H20D—C20'—H20E	109.5
N1—C12—H12B	109.5	C19'—C20'—H20F	109.5
H12A—C12—H12B	109.5	H20D—C20'—H20F	109.5
N1—C12—H12C	109.5	H20E—C20'—H20F	109.5
H12A—C12—H12C	109.5	C1—N1—C11	114.8 (2)
H12B—C12—H12C	109.5	C1—N1—C12	115.4 (2)
N2—C13—C14	112.71 (18)	C11—N1—C12	110.7 (2)
N2—C13—H13A	109.0	C17—N2—C13	119.76 (18)
C14—C13—H13A	109.0	C17—N2—S1	121.31 (15)
N2—C13—H13B	109.0	C13—N2—S1	118.44 (14)
C14—C13—H13B	109.0	C14—O4—C15	120.0 (4)
H13A—C13—H13B	107.8	C14—O4—C15'	105.2 (9)
O3—C14—O4	124.8 (2)	C18—O6—C19'	123.6 (6)
O3—C14—C13	125.6 (2)	C18—O6—C19	111.1 (4)
O4—C14—C13	109.5 (2)	O2—S1—O1	118.12 (10)
O4—C15—C16	107.0 (4)	O2—S1—N2	109.44 (10)
O4—C15—H15A	110.3	O1—S1—N2	106.18 (9)
C16—C15—H15A	110.3	O2—S1—C6	106.70 (9)
O4—C15—H15B	110.3	O1—S1—C6	111.09 (10)
C16—C15—H15B	110.3	N2—S1—C6	104.51 (9)
N1—C1—C2—C3	179.8 (2)	C18—C17—N2—C13	88.7 (2)
C10—C1—C2—C3	2.7 (4)	C18—C17—N2—S1	-83.1 (2)
C1—C2—C3—C4	0.7 (4)	C14—C13—N2—C17	81.6 (2)
C2—C3—C4—C5	-2.2 (4)	C14—C13—N2—S1	-106.41 (19)
C3—C4—C5—C10	0.2 (4)	O3—C14—O4—C15	8.1 (5)
C3—C4—C5—C6	-177.7 (2)	C13—C14—O4—C15	-172.2 (3)
C4—C5—C6—C7	175.7 (2)	O3—C14—O4—C15'	26 (2)
C10—C5—C6—C7	-2.3 (3)	C13—C14—O4—C15'	-155 (2)
C4—C5—C6—S1	-0.8 (3)	C16—C15—O4—C14	90.1 (10)
C10—C5—C6—S1	-178.75 (15)	C16—C15—O4—C15'	39 (4)
C5—C6—C7—C8	-1.0 (3)	C16'—C15'—O4—C14	163 (3)
S1—C6—C7—C8	175.69 (17)	C16'—C15'—O4—C15	-62 (3)
C6—C7—C8—C9	2.2 (3)	O5—C18—O6—C19'	-9.6 (5)
C7—C8—C9—C10	-0.2 (4)	C17—C18—O6—C19'	170.0 (4)
C8—C9—C10—C5	-3.2 (3)	O5—C18—O6—C19	11.4 (6)
C8—C9—C10—C1	179.9 (2)	C17—C18—O6—C19	-169.0 (6)

supplementary materials

C4—C5—C10—C9	-173.8 (2)	C20'—C19'—O6—C18	95.9 (12)
C6—C5—C10—C9	4.3 (3)	C20'—C19'—O6—C19	34.6 (15)
C4—C5—C10—C1	3.2 (3)	C20—C19—O6—C18	159.6 (9)
C6—C5—C10—C1	-178.75 (18)	C20—C19—O6—C19'	-72.0 (19)
C2—C1—C10—C9	172.3 (2)	C17—N2—S1—O2	126.11 (17)
N1—C1—C10—C9	-4.9 (3)	C13—N2—S1—O2	-45.80 (18)
C2—C1—C10—C5	-4.6 (3)	C17—N2—S1—O1	-2.41 (19)
N1—C1—C10—C5	178.17 (19)	C13—N2—S1—O1	-174.32 (15)
N2—C13—C14—O3	-10.0 (4)	C17—N2—S1—C6	-119.93 (17)
N2—C13—C14—O4	170.3 (2)	C13—N2—S1—C6	68.16 (17)
N2—C17—C18—O5	-10.7 (3)	C7—C6—S1—O2	14.37 (19)
N2—C17—C18—O6	169.75 (19)	C5—C6—S1—O2	-168.98 (16)
C2—C1—N1—C11	111.9 (3)	C7—C6—S1—O1	144.39 (16)
C10—C1—N1—C11	-71.0 (3)	C5—C6—S1—O1	-38.95 (19)
C2—C1—N1—C12	-18.7 (4)	C7—C6—S1—N2	-101.51 (17)
C10—C1—N1—C12	158.4 (2)	C5—C6—S1—N2	75.14 (17)

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C4—H4 \cdots O1	0.93	2.36	3.006 (3)	126
C13—H13A \cdots O2 ⁱ	0.97	2.36	3.272 (3)	157

Symmetry codes: (i) $-x+1, y-1/2, -z+1/2$.

Fig. 1

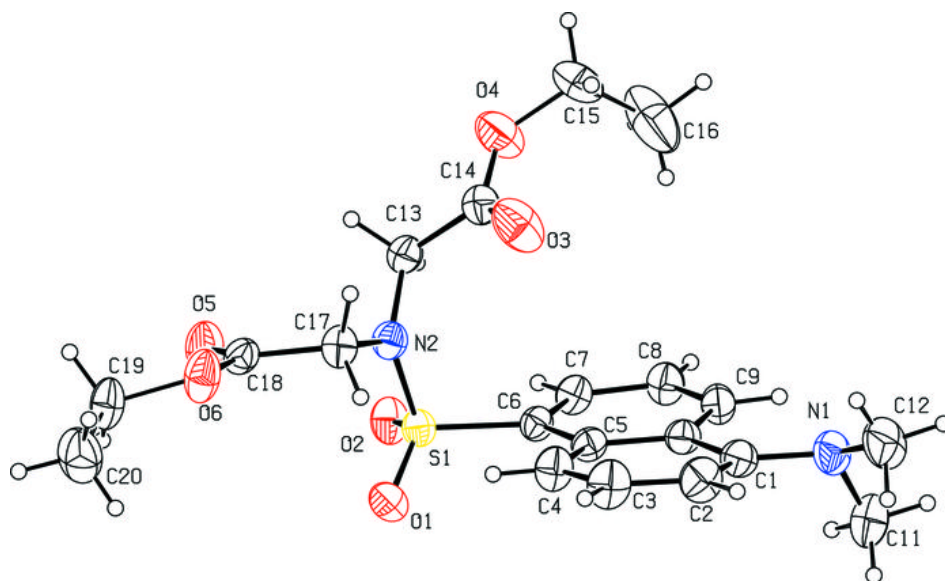


Fig. 2

